

# PHYSICAL SCIENCES JUNE 2014 EXAMPLER FOR P1 Study Guide

**Physical sciences June 2014 exemplar for P1** is a valuable resource for students preparing for their physics and chemistry exams. This exemplar provides insights into the types of questions, the exam format, and the key concepts that students need to master to excel in their assessments. Understanding the structure and content of the June 2014 P1 exam can significantly boost students' confidence and help them focus their revision on the most important topics. In this article, we will explore the main features of the June 2014 physical sciences exam, highlight common question types, and offer tips on how to approach such exams effectively.

## Overview of the Physical Sciences June 2014 Exemplar for P1

The June 2014 physical sciences exam was designed to test students' understanding of fundamental concepts in physics and chemistry. The exam typically consists of multiple-choice questions, short-answer questions, and longer problem-solving questions. The primary aim is to assess students' ability to apply theoretical knowledge to practical situations, interpret data, and perform calculations accurately.

## Exam Format and Structure

The June 2014 P1 exam followed a structured format including:

- Multiple-choice questions (MCQs): These test basic knowledge and quick recall.
- Short-answer questions: Designed to evaluate understanding of key concepts and ability to describe phenomena.
- Data interpretation questions: Require analysis of graphs, tables, and experimental data.
- Calculation questions: Involve applying formulas and performing mathematical operations to solve problems.

Understanding this structure helps students allocate their time efficiently during the exam.

## Key Topics Covered in the June 2014 P1 Exam

The exam covers a broad range of topics in physical sciences, with an emphasis on core concepts in physics and chemistry. Highlighted below are the main areas tested.

## Physics Topics

Physics questions in the exam often focus on:

- Forces and Motion: Understanding Newton's Laws, calculating speed, velocity, and acceleration.
- Energy and Work: Concepts of kinetic and potential energy, conservation of energy, and work done.
- Electricity: Basic principles of current, voltage, resistance, and simple circuits.
- Waves: Types of waves, properties, and wave behavior such as reflection and refraction.
- Sound and Light: Characteristics of sound waves and the behavior of light, including reflection and lenses.

## Chemistry Topics

Chemistry questions often involve:

- Atomic Structure: Understanding atoms, ions, and periodic table trends.
- Chemical Reactions: Types of reactions, balancing equations, and reaction rates.
- Acids and Bases: Properties, pH scale, and neutralization reactions.
- Materials and Their Properties: Metals, non-metals, and polymers.
- Environmental Chemistry: Pollution, renewable and non-renewable resources.

## Sample Questions from the June 2014 P1 Exam

To better understand what to expect, here are examples of typical questions from the exam:

### Physics Sample Question

**Q:** A car accelerates from a speed of 20 m/s to 30 m/s over a distance of 500 meters. Calculate the acceleration of the car.

**Solution:** Use the equation  $v^2 = u^2 + 2as$ . Rearrange to find  $a$ :

- $a = (v^2 - u^2) / 2s$
- $a = (30^2 - 20^2) / 2 \times 500$
- $a = (900 - 400) / 1000 = 500 / 1000 = 0.5 \text{ m/s}^2$

## Chemistry Sample Question

**Q:** Write the balanced chemical equation for the reaction between hydrochloric acid and sodium hydroxide.

**Answer:** The balanced equation is:



## Effective Strategies for Preparing for the June 2014 P1 Exam

Preparation is key to performing well in physical sciences exams. Here are some strategies tailored to the June 2014 P1 exam style.

### Understand the Exam Format

Knowing the structure of the exam helps in time management and reduces exam anxiety. Practice with past papers that mimic the June 2014 format to build familiarity.

### Focus on Core Concepts and Formulas

Most questions test understanding of fundamental principles and the ability to apply formulas. Create summary notes of key concepts, definitions, and equations.

### Practice Data Interpretation and Calculations

Work through past questions that involve analyzing graphs, tables, and performing calculations. This enhances problem-solving skills and accuracy.

### Use Past Exam Papers and Marking Schemes

Review previous exam papers, especially the June 2014 example, to identify common question patterns and mark schemes. This aids in understanding what examiners look for.

### Revise Laboratory and Practical Skills

Practical questions are common; ensure you are comfortable with basic laboratory techniques, safety procedures, and interpreting experimental data.

## Additional Resources for P1 Exam Preparation

To supplement your revision, consider the following resources:

- Past question papers and answer guides from previous years, especially the June 2014 paper.
- Revision guides and textbooks focusing on the P1 syllabus.
- Online tutorials and educational videos explaining key concepts.
- Study groups for collaborative learning and discussion of difficult topics.

## Conclusion

The **physical sciences June 2014 exemplar for P1** serves as an excellent benchmark for students aiming to excel in their physics and chemistry assessments. By understanding the exam format, practicing past questions, and mastering core concepts, students can improve their performance and build confidence. Remember, consistent revision, effective time management, and a clear understanding of fundamental principles are vital for success in physical sciences. Use the resources available, stay focused, and approach your exam with a positive mindset to achieve your academic goals.

### Physical Sciences June 2014 Examiner: An In-Depth Review and Analysis

The Physical Sciences June 2014 Examiner offers a comprehensive snapshot of the assessment standards, question design, and coverage expected at that examination session. For educators, students, and curriculum analysts alike, understanding the structure, content focus, and underlying assessment principles of this exam provides valuable insights into the pedagogical expectations and the competencies being tested. This review delves into the exam's framework, dissecting its components, evaluating question types, and offering critical analysis of its alignment with the curriculum objectives.

## Overview of the June 2014 Physical Sciences Exam

The June 2014 examination for Physical Sciences was structured to evaluate candidates across a broad spectrum of topics within physics and chemistry. The exam aimed to assess not only factual recall but also understanding, application of concepts, problem-solving skills, and analytical thinking.

The exam typically comprised two main sections:

- Section A: Multiple-choice questions designed to test foundational knowledge and basic comprehension.
- Section B: Structured questions requiring detailed explanations, calculations, and application-based responses.

This format aligns with the curriculum's emphasis on both factual knowledge and higher-order thinking skills.

## Content Coverage and Distribution

The exam content was distributed to reflect the key topics within the Physical Sciences syllabus for the grade level. These included:

### Physics Topics

- Motion and Forces
- Work, Energy, and Power
- Waves and Sound
- Light and Optics
- Electricity and Magnetism

### Chemistry Topics

- Atomic Structure and the Periodic Table
- Chemical Reactions and Equations
- Acids, Bases, and Salts
- Materials and Their Properties
- Chemical Bonding and Structure

The exam ensured a balanced coverage, with particular emphasis on areas deemed critical for foundational understanding.

## Question Structure and Types

A core feature of the June 2014 exam was its diverse question types designed to assess different cognitive levels.

### Multiple-Choice Section

- Consisted of 15-20 questions
- Focused on recall, recognition, and straightforward application
- Designed for quick assessment of core concepts

## Structured Questions

- Included 5-7 comprehensive questions
- Required detailed written responses
- Incorporated calculations, diagram drawing, and explanation
- Assessed application and analytical skills

## Practical and Data-Handling Questions

- Some questions involved interpreting experimental data
- Required understanding of experimental procedures and results
- Tested skills in data analysis, graph plotting, and conclusion drawing

## Analysis of Key Questions and Topics

In-depth analysis of select questions provides insight into the exam's standards and expectations.

### Physics: Motion and Forces

A typical question asked candidates to:

- Calculate the acceleration of an object given initial velocity, final velocity, and time.
- Explain the relationship between force, mass, and acceleration using Newton's second law.
- Draw free-body diagrams to illustrate forces acting on objects.

This assessed both quantitative skills and conceptual understanding.

### Chemistry: Atomic Structure

Questions here involved:

- Identifying isotopes based on atomic and mass numbers.
- Explaining the arrangement of electrons in atoms.
- Describing the periodic trends such as atomic radius and electronegativity.

Such questions gauged students' grasp of fundamental chemical principles.

## Application and Problem-Solving

An example involved:

- Calculating the energy transferred in a heating process.
- Explaining the practical implications of electrical resistance in circuits.

This demonstrated the exam's emphasis on applying theoretical knowledge to real-world scenarios.

## Assessment Standards and Marking Criteria

The exam adhered to strict marking schemes to ensure fairness and consistency:

- Clear rubrics for each question, specifying points for correct calculations, explanations, and diagrams.
- Emphasis on showing steps in calculations for partial credit.
- Awarding marks for clarity, accuracy, and completeness.

The marking criteria aimed to reward both correct knowledge and logical reasoning processes, encouraging students to demonstrate their understanding comprehensively.

## Strengths of the June 2014 Examination

- **Balanced Coverage:** The exam effectively covered a wide range of topics, ensuring a holistic assessment.
- **Diverse Question Formats:** The inclusion of multiple-choice, structured, and data-based questions catered to different learning styles and cognitive skills.
- **Application Focus:** Questions linked theory to practical contexts, fostering real-world understanding.
- **Clear Marking Schemes:** Well-defined rubrics facilitated objective grading and transparency.

## Challenges and Areas for Improvement

While the exam was robust, certain areas could be refined:

- **Question Clarity:** Some questions, particularly in physics, could have been more precisely worded to avoid ambiguity.

- Difficulty Level Balance: A few questions appeared disproportionately challenging, potentially disadvantaging less-prepared candidates.
- Inclusion of Contemporary Contexts: The exam could integrate more current scientific developments to enhance relevance and engagement.

## Implications for Teaching and Preparation

Educators can leverage insights from the June 2014 exam to tailor instruction:

- Emphasize both conceptual understanding and calculation skills.
- Use past questions to familiarize students with exam formats and expectations.
- Incorporate practical and data analysis exercises in teaching plans.
- Focus on areas identified as high-yield topics, such as forces, atomic structure, and electrical circuits.

Students should practice a variety of question types, develop clear explanations, and refine their problem-solving strategies to excel in future assessments.

## Conclusion

The Physical Sciences June 2014 Examiner exemplifies a well-structured, comprehensive assessment aligned with curriculum objectives, designed to evaluate a broad spectrum of scientific competencies. Its balanced approach to testing foundational knowledge, application skills, and analytical thinking provides meaningful insights into the educational standards of that examination period.

Analyzing this exam offers valuable lessons for educators seeking to improve instructional strategies and for students aiming to optimize their exam preparation. As scientific education continues to evolve, future assessments can build on such frameworks, integrating contemporary themes and innovative question formats to better prepare learners for the challenges of modern science.

In Summary:

- The June 2014 exam balanced breadth and depth across physics and chemistry topics.
- It employed varied question types to assess diverse skills.
- Its structure and marking schemes promoted fairness and clarity.
- Continuous reflection on such exams informs curriculum development and teaching methodologies.



This detailed review underscores the importance of meticulous exam design and the role of high-quality assessments in fostering scientific literacy.

**Question** What are the key topics covered in the June 2014 Physical Sciences P1 exam? The June 2014 Physical Sciences P1 exam primarily covered topics such as mechanics, properties of matter, and basic electricity, focusing on fundamental concepts and calculations related to these areas. **Answer** How can I effectively prepare for the Physical Sciences P1 June 2014 exam? Effective preparation involves reviewing past exam papers, understanding core concepts, practicing calculations, and solving sample questions to familiarize yourself with the exam format and common question types. What are some common types of questions asked in the June 2014 P1 exam? Common questions include calculations involving speed, velocity, and acceleration, explanations of properties of materials, and basic electrical circuit analysis based on the June 2014 exam. Are there any specific formulas I should memorize for the June 2014 P1 exam? Yes, key formulas include equations of motion, Ohm's law, and formulas relating to pressure, density, and the properties of materials, which are essential for solving many questions in the exam. What strategies can I use to improve my performance in the mechanics section of the June 2014 P1 exam? Focus on understanding the concepts of force, motion, and energy, practice applying formulas to different scenarios, and work through past questions to improve problem-solving speed and accuracy. How important are diagram drawing and labeling in the June 2014 P1 exam? Drawing clear, labeled diagrams helps in visualizing problems, understanding the question better, and often earns partial credit, making it a crucial skill for the exam. What are some tips for managing time effectively during the June 2014 P1 exam? Allocate specific time blocks for each section or question, start with questions you are confident in, and leave difficult questions for later to ensure you complete the paper within the given time. How can I interpret and analyze data or graphs in the June 2014 P1 exam? Carefully read axis labels, note units, identify trends or patterns, and use the data to answer related questions accurately, ensuring a thorough understanding of the graphical information. Where can I find official past papers and marking schemes for the June 2014 Physical Sciences P1 exam? Official past papers and marking schemes are available on the examination board's website or through your school's resource center, providing valuable practice and insight into exam expectations.

**Related keywords:** physics, chemistry, biology, exam preparation, past papers, question paper, sample exam, science exam, June 2014, P1 module

## physical science creative teaching press physical changes

**Physical science creative teaching press physical changes** is a vital topic in the realm of science education, especially for educators aiming to foster understanding and curiosity about the

fundamental nature of matter. Understanding physical changes is essential for students to grasp how substances transform without altering their chemical composition. This article explores the concept of physical changes in physical science, emphasizing creative teaching methods, key principles, real-world examples, and effective strategies to make learning engaging and comprehensive.

## Understanding Physical Changes in Physical Science

### What Are Physical Changes?

Physical changes refer to alterations in the form, appearance, or state of a substance that do not modify its chemical identity. These changes are often reversible, meaning the original substance can typically be recovered without chemical reactions taking place.

Examples of physical changes include:

- Melting ice into water
- Boiling water into vapor
- Cutting paper
- Dissolving sugar in tea
- Crushing a can

Key characteristics of physical changes:

- No new substances formed
- Usually reversible
- Involve changes in physical properties like shape, size, or phase

### Difference Between Physical and Chemical Changes

While physical changes involve no change in chemical composition, chemical changes result in the formation of new substances with different properties.

| Aspect     | Physical Change                  | Chemical Change             |
|------------|----------------------------------|-----------------------------|
| -----      | -----                            | -----                       |
| Definition | Change in physical form or state | Formation of new substances |

| Reversibility | Usually reversible | Usually irreversible |

| Example | Melting ice | Burning wood |

## Creative Teaching Strategies for Physical Changes

Engaging students in understanding physical changes requires innovative and creative teaching methods. Here are some effective strategies:

### Hands-On Experiments and Demonstrations

Practical activities help students visualize and understand physical changes concretely.

Examples include:

- Melting ice cubes and observing the change
- Evaporating saltwater
- Crumpling aluminum foil
- Freezing water to form ice
- Dissolving sugar and salt

Tips for effective experiments:

- Use clear, simple materials
- Encourage students to record observations
- Discuss reversible vs. irreversible changes

### Interactive Visual Aids and Simulations

Utilize multimedia tools like animations and virtual labs to demonstrate physical changes that are difficult to observe directly.

Benefits include:

- Enhancing visualization skills
- Providing safe environments for experimentation
- Catering to diverse learning styles

## Storytelling and Real-Life Contexts

Integrate stories or real-world examples to make the concept relatable.

Examples:

- How ice melts outside on a sunny day
- How a puddle evaporates after rain
- The process of freezing water into ice cubes at home

## Creative Classroom Activities

Engage students through activities that promote exploration and discussion.

Ideas include:

- "Change Detective" game where students identify physical changes around them
- Group projects on different physical changes observed in nature
- Art projects like sculpting or paper folding to demonstrate physical transformations

## Key Concepts and Principles of Physical Changes

### States of Matter and Phase Changes

Understanding how matter transitions between solid, liquid, and gas states is central to physical science.

Common phase changes:

- Melting (solid to liquid)
- Freezing (liquid to solid)
- Vaporization or boiling (liquid to gas)
- Condensation (gas to liquid)
- Sublimation (solid to gas)

Teaching tip: Use diagrams and models to illustrate these transitions dynamically.

### Properties of Materials During Physical Changes

Monitor how physical properties such as shape, volume, density, and color change during physical transformations.

Examples:

- Volume expansion of metal when heated
- Changes in shape during molding or crushing
- Color changes due to temperature effects

## **Reversibility and Conservation of Mass**

Most physical changes are reversible and adhere to the law of conservation of mass, meaning the total mass remains constant during the change.

Discussion points:

- How physical changes do not alter the chemical identity
- Examples demonstrating conservation of mass in physical changes

## **Real-World Applications of Physical Changes**

Understanding physical changes has numerous applications across industries and daily life.

### **In Manufacturing and Industry**

- Metal forging and shaping
- Plastic molding
- Food processing (freezing, boiling, drying)

### **In Nature and Environment**

- Water cycle processes like evaporation and condensation
- Glacial melting
- Weathering of rocks

### **In Daily Life**

- Cooking (boiling, freezing)
- Cleaning (dissolving detergents)
- Recycling materials like aluminum cans

## Assessment and Evaluation Strategies

To gauge understanding, educators can implement various assessment tools:

- Quizzes and Worksheets: Focused on identifying physical changes
- Observation and Lab Reports: Documenting experiments
- Group Discussions: Explaining processes and concepts
- Creative Projects: Designing models or presentations illustrating physical changes

## Challenges and Solutions in Teaching Physical Changes

Common challenges include:

- Abstract concepts difficult for students to grasp
- Confusing physical changes with chemical changes
- Limited access to laboratory resources

Solutions:

- Use visual aids and models extensively
- Reinforce concepts with real-life examples
- Incorporate virtual labs and simulations
- Encourage critical thinking and questioning

## Conclusion

Teaching physical science with a focus on physical changes demands creativity, engagement, and clarity. By employing hands-on experiments, visual tools, storytelling, and real-world contexts, educators can foster a deep understanding of physical changes among students. Recognizing the importance of these concepts not only enhances scientific literacy but also prepares learners to appreciate the dynamic nature of matter in everyday life and various industries. As educators continue to innovate in their teaching approaches, the topic of physical changes remains a fundamental and exciting area of physical science education.

Keywords for SEO Optimization:

- Physical science
- Creative teaching methods
- Physical changes
- Science education
- Phase changes
- Reversible changes
- Hands-on science experiments
- Teaching physical science effectively
- Real-world science applications

## Physical Science Creative Teaching Press Physical Changes: A Comprehensive Guide to Engaging Science Education

In the realm of physical science education, understanding the concept of physical changes is fundamental. When educators utilize resources from the Creative Teaching Press, particularly their materials on physical science, they open doors to innovative, hands-on learning experiences. This guide delves into the essence of physical changes, exploring how teachers can effectively introduce, demonstrate, and assess this critical scientific concept using creative teaching strategies and resources.

### Understanding Physical Changes in Science

Physical changes refer to alterations in the state or appearance of a substance without changing its chemical composition. These changes are often reversible and are fundamental to students' grasp of material properties, states of matter, and the nature of physical phenomena.

### Key Characteristics of Physical Changes:

- No new substance forms
- Usually reversible
- Changes involve physical properties like shape, size, phase, or texture
- Do not affect the chemical identity of the substance

Examples include melting, freezing, boiling, condensation, breaking, and dissolving.

### Why Focus on Physical Changes?

Teaching about physical changes lays the groundwork for understanding more complex topics like chemical reactions and molecular structure. It also encourages observational skills, critical thinking, and science process skills such as hypothesizing, experimenting, and analyzing.

Using creative teaching methods, including hands-on activities, visual aids, and real-world applications, enhances student engagement and comprehension.

### Effective Strategies for Teaching Physical Changes

#### - Hands-On Experiments and Demonstrations

Experiments are the cornerstone of science teaching. Engaging students with tangible experiences helps solidify abstract concepts.

#### Sample Activities:

- Melting Ice: Show how ice melts into water when heated, emphasizing phase change from solid to liquid.
- Dissolving Salt or Sugar: Demonstrate solubility and how substances can undergo physical changes without chemical reactions.
- Crushing or Breaking: Break chalk or plastic to illustrate physical alteration, emphasizing that the material remains the same.

#### Tips:

- Use clear, simple materials.
- Encourage students to record observations and hypothesize outcomes.
- Discuss reversibility after each activity.

#### - Visual Aids and Graphic Organizers

Visual tools help students visualize the concept and organize their understanding.

#### Examples:

- Charts comparing physical and chemical changes



- Diagrams illustrating phase changes
- Flowcharts depicting the steps in various physical processes
- Incorporate Creative Media and Resources

Creative Teaching Press offers a suite of resources, including posters, activity books, and digital content, designed to make learning about physical changes interactive and memorable.

Utilize:

- Illustrated posters showing different physical changes
- Interactive worksheets that challenge students to classify changes
- Story-based scenarios that involve physical change concepts
- Real-Life Applications and Contextual Learning

Connecting science to students' everyday experiences makes lessons relevant and meaningful.

Examples:

- Melting ice in a beverage
- Freezing water to make ice cubes
- Dissolving sugar in coffee or tea
- Physical alterations in manufacturing or cooking

### Classroom Activities and Lesson Plans

Here's a sample outline for a comprehensive lesson:

Lesson Title: Exploring Physical Changes

Objectives:

- Define physical change
- Identify different types of physical changes
- Conduct simple experiments to observe physical changes

**Materials Needed:**

- Ice cubes
- Water
- Salt
- Sugar
- Plastic or glass containers
- Crayons or chalk
- Safety goggles

**Procedure:**

- Introduction (10 minutes): Discuss what physical changes are, using visuals.
- Demonstration (15 minutes): Melting ice and dissolving salt/sugar.
- Student Activity (30 minutes): Students perform experiments:

- Observe ice melting
- Dissolve sugar in water
- Break chalk into pieces
- Freeze water and observe ice formation

- Discussion (15 minutes): Students share observations, compare results, and classify each change.

- Assessment (10 minutes): Quiz or worksheet to reinforce understanding.

**Assessment Tips:**

- Use questioning strategies to assess understanding.
- Have students draw diagrams of their experiments and label physical changes.

**Assessing Understanding of Physical Changes**

Assessment should gauge both conceptual understanding and practical skills.

**Methods:**

- Observation during activities
- Student reflections and explanations
- Quizzes with multiple-choice and short-answer questions
- Concept maps illustrating types of changes

### Common Misconceptions and How to Address Them

- Misconception: Physical changes always involve a change in the substance's chemical identity.

Correction: Emphasize that physical changes do not alter the chemical structure, using examples like melting ice or tearing paper.

- Misconception: All physical changes are reversible.

Correction: Clarify that some physical changes, like breaking or cutting, are reversible, but others, like tearing or dissolving, may not be easily reversible.

### Integrating Creative Teaching Press Resources

Creative Teaching Press provides various materials tailored for physical science instruction, which can elevate your teaching approach:

- Activity Books: Ready-made experiments and activities that align with standards.
- Posters and Visuals: Engaging images that reinforce concepts.
- Interactive Worksheets: Opportunities for students to classify and analyze physical changes.
- Digital Resources: Online games and quizzes to supplement lessons.

### Strategies for Integration:

- Use posters as classroom decor to continually reinforce concepts.
- Incorporate activity books into small-group work or centers.
- Assign digital quizzes as formative assessments.

### Extending Learning Beyond the Classroom

Encourage students to observe physical changes in their environment.

## Assignments:

- Keep a science journal recording physical changes observed at home or outdoors.
- Research and present on real-world applications of physical changes in industries.
- Design their own experiments related to physical changes and share findings.

## Conclusion

Understanding physical changes is a vital component of physical science education. By leveraging creative teaching strategies and resources, such as those provided by Creative Teaching Press, educators can foster curiosity, deepen understanding, and nurture scientific literacy among students. From hands-on experiments to visual aids and real-world connections, teaching physical changes effectively prepares students for more advanced topics and encourages a lifelong appreciation for science.

Remember: The goal is to make science accessible, engaging, and relevant. Through creative teaching approaches, educators can inspire the next generation of scientists and thinkers to explore the fascinating world of physical science.

**Question** What are some effective creative teaching strategies for explaining physical changes in science classes? **Answer** Creative strategies include hands-on experiments, visual demonstrations, interactive models, and storytelling to help students grasp concepts like melting, freezing, and condensation in engaging ways. How can teachers make the concept of physical changes more relatable to students? Teachers can use everyday examples such as ice melting, boiling water, or folding paper to illustrate physical changes, making the concepts more familiar and easier to understand. What role does the Physical Science Creative Teaching Press play in improving science education? The Physical Science Creative Teaching Press provides innovative resources, lesson plans, and activities designed to foster creativity and enhance student understanding of physical science concepts like physical changes. What are some common misconceptions students have about physical changes, and how can creative teaching address them? Students often confuse physical changes with chemical changes. Creative teaching methods like experiments and visual aids can clarify that physical changes do not alter the substance's identity, helping to correct misconceptions. Can technology enhance teaching about physical changes in physical science? If so, how? Yes, technology such as virtual labs, simulations, and interactive videos can provide dynamic and visual representations of physical changes, making abstract concepts more accessible and engaging for students.

**Related keywords:** physical science, creative teaching, physical changes, science education, hands-on experiments, teaching resources, science curriculum, classroom activities, scientific concepts, educational publishing

**Read the full article online:** [physical science creative teaching press physical changes](#)